# Tech Notes

## **Conversion Factors For Emissions Calculations**

Preparing emissions estimates for environmental authorities can be difficult because they often ask for emissions expressed in units not available through existing data. Here are the conversion procedures for some of the more commonly-used measurement systems:

#### ppm at 3% O<sub>2</sub> (15% excess air) in dry flue gases to lb./million Btu (ppm)(F<sub>3</sub>) = lb./million Btu

Various Fuels	NOx Measured As NO <sub>2</sub>	со	Aldehydes, Measured As Formaldehyde	Unburned Hy Measu Methane	/drocarbons, red As: Propane	CO <sub>2</sub>	SO <sub>2</sub>
Birmingham Nat. Gas*	.001187	.000722	.000781	.000416	.001147	.001147	.001672
Propane	.001185	.000721	.000780	.000415	.001146	.001146	.001669
Butane	.001212	.000735	.000798	.000424	.001172	.001172	.001707
#2 Oil**	.001317	.000801	.000867	.000461	.001273	.001273	.001854

Values of multiplier F3 for various fuels and emissions

#### lb./million Btu to ppm at 3% O<sub>2</sub> (15% excess air) in dry flue gases

(lb./million Btu)(f<sub>3</sub>) = ppm @ 3% O<sub>2</sub>, dry

Values of multiplier f3 for various fuels and emissions

Various Fuels	NOx Measured As NO <sub>2</sub>	со	Aldehydes, Measured As Formaldehyde	Unburned Hy Measu Methane		CO <sub>2</sub>	SO <sub>2</sub>
Birmingham Nat. Gas*	842	1385	1280	2404	872	872	598
Propane	844	1387	1282	2410	873	873	599
Butane	825	1361	1253	2358	853	853	586
#2 Oil**	759	1248	1153	2169	786	786	539

#### 3) ppm at 0% O<sub>2</sub> in dry flue gases to lb./million Btu (ppm)(F<sub>0</sub>) = lb./million Btu

#### Values of multiplier Fo for various fuels and emissions

Various Fuels	NOx Measured As NO <sub>2</sub>	со	Aldehydes, Measured As Formaldehyde	Unburned Hy Measu Methane	· · ·	CO <sub>2</sub>	SO <sub>2</sub>
Birmingham Nat. Gas*	.001017	.000617	.00067	.000356	.000983	.000983	.001432
Propane	.001018	.000619	.00067	.000356	.000984	.000984	.001434
Butane	.001042	.000634	.000686	.000365	.001007	.001007	.001468
#2 Oil**	.001133	.00069	.000746	.000397	.001096	.001096	.001596

\* 1002 Gross Btu/cubic foot, 8.48 Cubic feet dry flue products at stoichiometric ratio.

 $^{\star\star}$  Calculated as heptadecane,  $\rm C_{_{17}}\,H_{_{36}},\,19,270$  Gross Btu/lb.

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#### 4) **lb./million Btu to ppm at 0% O<sub>2</sub> in dry flue gases** (lb./million Btu)(f<sub>0</sub>) = ppm @ 0% O<sub>2</sub>, dry

Various Fuels	NOx Measured As NO	со	Aldehydes, Measured As Formaldehyde	Unburned Hy Measur Methane		CO <sub>2</sub>	SO <sub>2</sub>
Birmingham Nat. Gas*	983	1621	1493	2809	1017	1017	698
Propane	982	1616	1493	2809	1016	1016	697
Butane	960	1577	1458	2740	983	983	681
#2 Oil**	883	1449	1340	2519	912	912	627

Values of multiplier fo for various fuels and emissions

#### 5) ppm at 3% $O_2$ or 0% $O_2$ in dry flue gases to lb./year

First, calculate lb./million Btu with Step 1 or 3 on the first page. Then convert to lbs./year with the following relationship:

(lb./million Btu) (Maximum Burner Input, million Btu/hr.) (operating hrs./year) = lb./year

### 6) lb/year to ppm at 3% $O_2$ or 0% $O_2$ in dry flue gases

lb./year ÷ operating hrs./year ÷ Maximum Burner Input, million Btu/hr. = lb./million Btu

Convert lb./million Btu to ppm with Step 2 or 4.

#### 7) ppm at 3% O<sub>2</sub> or 0% O<sub>2</sub> in dry flue gases to gm/Nm<sup>3</sup> (ppm)(G) = gm/Nm<sup>3</sup>

 $\sin(\alpha) = gm/m^{2}$ 

	NOx		Aldehydes,	Unburned Hy	drocarbons,		
	Measured		Measured As	Measu	ed As:		
Emission	As NO <sub>2</sub>	CO	Formaldehyde	Methane	Propane	CO <sub>2</sub>	SO <sub>2</sub>
G	.002031	.001235	.001341	.000716	.001969	.001965	.002861

Values of multiplier G for various emissions

## 8) **gm/Nm<sup>3</sup> to ppm at 3% O<sub>2</sub> or 0% O<sub>2</sub> in dry flue gases** (gm/Nm<sup>3</sup>)(g) = ppm

Values of multiplier	g fo	r various	emissions
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Emission	NOx Measured As NO <sub>2</sub>	со	Aldehydes, Measured As Formaldehyde	Unburned Hy Measur Methane	,	CO <sub>2</sub>	SO2
g	492.4	809.7	745.7	1396.6	507.9	508.9	349.5

\* 1002 Gross Btu/cubic foot, 8.48 Cubic feet dry flue products at stoichiometric ratio.

\*\* Calculated as heptadecane, C<sub>17</sub>H<sub>36</sub>, 19,270 Gross Btu/lb.

# Tech Notes

## Correcting Emissions Readings to 3% O<sub>2</sub> or 11% O<sub>2</sub> Basis

Many environmental authorities, including the U.S. EPA and several European agencies, require that gaseous pollutants, like  $NO_x$  and CO, be reported in ppm (parts per million by volume) corrected to a based of 3% excess  $O_2$ —or 15% excess air—in the flue gases. Japan, on the other hand, customarily uses a base of 11%  $O_2$ .

Emission readings taken at different oxygen levels can be easily converted to a standard base using a multiplier:

### $ppm_{corrected} = ppm_{test} x multiplier$

The multiplier is calculated from the oxygen reading taken during the test and the base oxygen reading required by the regulation:

multiplier = 
$$\frac{21 - \% O_2 \text{ base}}{21 - \% O_2 \text{ test}}$$

For your convenience, a table of multipliers is presented to the right.

	Multiplier For:					
%O <sub>2</sub>	3%O <sub>2</sub>	11%0 <sub>2</sub>				
0	.86	.48				
1	.9	.5				
2	.95	.53				
3	1	.56				
4	1.06	.59				
5	1.13	.63				
6	1.2	.67				
7	1.29	.71				
8	1.38	.77				
9	1.5	.83				
10	1.64	.91				
11	1.8	1				
12	2.0	1.11				
13	2.25	1.25				
14	2.57	1.43				
15	3.0	1.67				
16	3.6	2				
17	4.5	2.5				
18	6	3.33				
18.5	7.2	4				
19	9	5				
19.5	12	6.67				
20	18	10				
20.2	22.5	12.5				
20.4	30	16.67				
20.6	45	25				
20.8	90	50				